

DITERPENOID QUINONES OF *SALVIA LANATA**

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Key Word Index—*Salvia lanata*, Labiatae, 20-hydroxy-7 α -acetoxyroyleanone, royleanone

Abstract—Two diterpenoid quinones, 20-hydroxy-7 α -acetoxyroyleanone and royleanone, have been isolated from the whole plant (aerial parts and roots) of *Salvia lanata*.

In our previous communications [1, 2], we reported the isolation of three diterpenoid quinones, horminone, decacylnemorone and 7 α -acetoxyroyleanone, and a new pentacyclic triterpenic acid, 3-*epi*-ursolic acid, from the petrol extract of the whole plant (aerial parts and roots) of *Salvia lanata* Roxb. In this communication we report the isolation of two more diterpenoids having an abietane skeleton from the benzene extract of the plant. One of them is royleanone whilst the other appears to be a new quinone, the structure of which has been elucidated as 20-hydroxy-7 α -acetoxyroyleanone (**1**) on the basis of chemical and spectral studies.

20-Hydroxy-7 α -acetoxyroyleanone, C₂₂H₃₀O₆ (M⁺, *m/z* 390), mp 160–162° exhibits UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm(ϵ) 280 (12 000) and 407 (600). Its IR spectrum shows the presence of acetyl, hydrogen bonded hydroxyl and α,β -unsaturated carbonyl groups. Its ¹H NMR spectrum (CDCl₃, 100 MHz) is strikingly similar to that of 7 α -acetoxyroyleanone (**3**) [3–5] but lacks the angular methyl proton signal at δ 1.30. Instead, it shows a peak at δ 3.58 (2H, *d*, *J* = 10 Hz) indicating the presence of a –CH₂OH group. The presence of two intense peaks at *m/z* 359 [M – CH₂OH]⁺ and 344 [M – CH₂OH – Me]⁺ in the mass spectrum of the compound also supports the above view. Further evidence in favour of the structure **1** for the diterpenoid is provided by the nature of the unsaponifiable product of this compound. Thus, saponification of **1** gave rise to **2**, C₂₀H₂₈O₅, mp 165–168°, the ¹H NMR spectrum (CDCl₃, 100 MHz) of which, as expected, is practically identical to that of horminone (**4**) [1, 4, 5], except for the absence of an angular methyl proton signal. Instead, it has a doublet (2H, *J* = 10 Hz) around δ 3.64 for

a –CH₂OH group. All these spectral and chemical observations reveal that the diterpenoid differs from 7 α -acetoxyroyleanone (**3**) by replacement of the methyl group at C-10 with a hydroxymethyl group. Its structure is, therefore, 20-hydroxy-7 α -acetoxyroyleanone (**1**).

EXPERIMENTAL

All mps are uncorr.

Isolation of royleanone and 1 Ca 1 kg of defatted plant material was extracted with C₆H₆ in a Soxhlet for 45 hr. The extract was subjected to CC on 200 g Si gel (mesh 60–120). The petrol (60–80)–C₆H₆ (1:1) eluate yielded royleanone which was crystallized from MeOH as yellow plates. C₂₀H₂₈O₃ (M⁺, *m/z* 316), mp 182°, yield 0.08 g. Its spectral data were similar to those reported in the lit. [3].

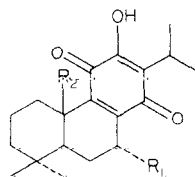
The fraction eluted with C₆H₆ gave 20-hydroxy-7 α -acetoxyroyleanone. It crystallized as brown plates from CHCl₃–*n*-hexane (1:3), yield 0.4 g, mp 160–162°. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm(ϵ) 280 (12 000) and 407 (600). IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1} 3440 (OH), 1735, 1665 and 1625. ¹H NMR (CDCl₃, 100 MHz) δ 0.90 (6H, *s*, 2 \times Me), 1.10 and 1.15 (6H, *d*, *J* = 6 Hz, –CHMe₂), 2.05 (3H, *s*, OOC Me), 3.25 (1H, *septet*, *J* = 7 Hz, –CHMe₂), 3.60 (2H, *d*, *J* = 10 Hz, –CH₂OH), 5.70 (1H, *br*, *W*₁ = 8 Hz, –CH=O–CO Me) and 7.20 (1H, *br*, *s*, phenolic–OH).

Saponification of 1 20-Hydroxy-7 α -acetoxyroyleanone (**1**, 0.03 g) in MeOH (5 ml) was refluxed for 4 hr with 10 ml 10% methanolic KOH. The saponification mixture was then cooled, filtered, acidified with dil. HCl and extracted with CHCl₃. The CHCl₃ extract was then washed with H₂O, dried over Na₂SO₄ and the solvent removed. On repeated CC over Si gel (50 g) the residue furnished **2**, C₂₀H₂₈O₅, mp 165–168°. IR $\nu_{\text{max}}^{\text{KBr}}$ cm^{–1} 3510, 3340, 1665 and 1625. ¹H NMR (CDCl₃, 100 MHz) δ 0.90 (6H, *s*, 2 \times Me), 1.10 and 1.15 (6H, *d*, *J* = 6 Hz, –CHMe₂), 3.25 (1H, *septet*, *J* = 7 Hz, –CHMe₂), 3.64 (2H, *d*, *J* = 10 Hz, –CH₂OH), 4.75 (1H, *br*, *s*, *W*₁ = 8 Hz, –CH(OH)–) and 7.20 (1H, *br*, *s*, phenolic–OH).

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REFERENCES

1. Mukherjee, K. S., Ghosh, P. K. and Sub. Badaruddin, (1981) *Phytochemistry* **20**, 1441.



1. R₁ = OAc, R₂ = CH₂OH
2. R₁ = OH, R₂ = CH₂OH
3. R = OAc, R₂ = Me
4. R₁ = OH, R₂ = Me

*Part 2 in the series "Diterpenoid Quinones of *Salvia lanata*"
 For Part 1 see ref. [1].

2. Mukherjee, K. S., Bhattacharya, M. K. and Ghosh, P. K. (1982) *Phytochemistry* **21**, 2416
3. Edwards, O. E., Feniak, G. and Los, M. (1962) *Can. J. Chem.* **40**, 1540
4. Brieskorn, C. H. and Buchberger, L. (1973) *Planta Med.* **24**, 190
5. Yoshizaki, F., Ruedi, P. and Eugster, C. H. (1979) *Helv. Chim. Acta* **62**, 2754

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LIMONOID EXTRACTIVES FROM *XYLOCARPUS MOLUCCENSIS*

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Key Word Index—*Xylocarpus moluccensis*, Meliaceae; limonoids; phragmalin, xylocensin G, xylocensin H

Abstract—Further examination of the timber of *Xylocarpus moluccensis* has given three new compounds, xylocensins G, H and I. Structures are deduced for G and H. A structure is also deduced for xylocensin C and proposals are made concerning the biosynthesis of phragmalin

The outstanding problem in the phytochemistry of the limonoids is how the ring A bridge is formed in the 1,29-cycloswietenan group of compounds, of which phragmalin (1) is the simplest known example. So far the only

plant investigated containing cycloswietenans and simpler limonoids which appear possible as biosynthetic intermediates is *Xylocarpus moluccensis* [1], which has given phragmalin triacetate (xylocensin E), a group of 1,8-

